

Rhenium-catalyzed Regio- and Stereoselective Dimerization and Cyclotrimerization of Terminal Alkynes

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A combination of catalytic amounts of $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ and tetrabutylammonium fluoride (TBAF) promoted the dimerization of aryl- and alkenyl-substituted alkynes with high regio- and stereoselectivities, and gave only (*E*)-enynes. When terminal alkynes having a carbonyl group were employed as substrates, regioselective cyclotrimerization of the alkynes occurred to give 1,3,5-trisubstituted aromatic compounds.

Regio- and stereoselective dimerization and cyclotrimerization of terminal alkynes are efficient and useful methods to synthesize enynes and trisubstituted aromatic compounds. Therefore, there have been many reports on dimerization¹ and cyclotrimerization^{2,3} of terminal alkynes. However, a mixture of (*E*)-1,4-disubstituted-, (*Z*)-1,4-disubstituted-, and 2,4-disubstituted-enynes from dimerization (Figure 1a) or a mixture of 1,3,5- and 1,2,4-trisubstituted aromatic compounds from trimerization (Figure 1b) is usually formed. Therefore, it is difficult to construct (*E*)-enyne and 1,3,5-tricarbonyl-substituted benzene skeletons regio- and stereoselectively. We report herein the regio- and stereoselective synthesis of (*E*)-enynes and 1,3,5-tricarbonyl-substituted benzenes from terminal alkynes using a mixture of a rhenium complex and tetrabutylammonium fluoride (TBAF) as a catalyst.

During our investigations of rhenium-catalyzed transformations using alkynes,⁴ we found that the dimerization of terminal alkynes took place in the presence of a rhenium complex, $\text{Re}_2(\text{CO})_{10}$, and TBAF as a catalyst. For example, treatment of phenylacetylene (**1a**) with catalytic amounts of the combination in toluene at 80 °C for 8 h gave (*E*)-enyne **2a** in 50% yield (eq 1).^{5,6} In this reaction, (*Z*)-1,4-diphenyl-1-buten-3-yne and 2,4-diphenyl-1-buten-3-yne, which are stereo- and regioisomers of **2a**, were not generated at all. In addition, cyclotrimerization, oligomerization, and polymerization of phenylacetylene (**1a**) did not occur.

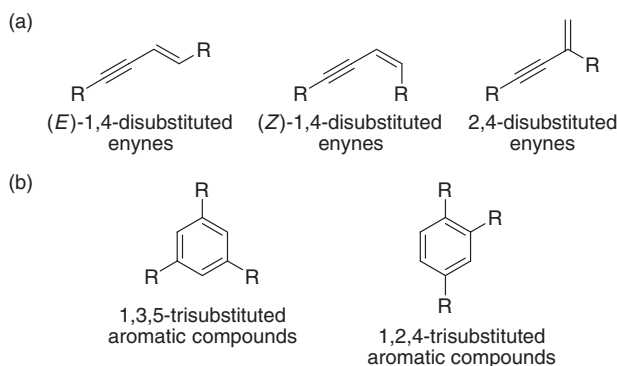


Figure 1. Isomers of dimerization and cyclotrimerization products of terminal alkynes.

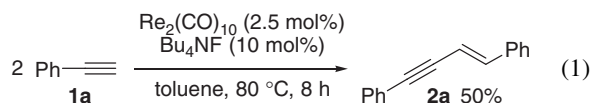


Table 1. Investigation of several rhenium complexes and additives

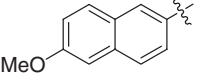
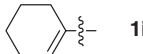
Entry	Catalyst	Additive	Yield/% ^a
1	$\text{ReBr}(\text{CO})_5^b$	Bu_4NF	6
2	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	Bu_4NF	98
3	—	Bu_4NF	0
4	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	—	0
5	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	Bu_4NF^b	4
6	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	DMA	0
7	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	DBU	10
8	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	<i>i</i> -Pr ₂ NEt	0

^a ¹H NMR yield. ^b 5.0 mol %.

To improve the yield of (*E*)-enyne **2a**, several catalysts and additives were investigated (Table 1). When a rhenium complex, $\text{ReBr}(\text{CO})_5$, was used, **2a** was formed only in 6% yield (Entry 1). By replacing some of the carbonyl groups with THF ligands, the yield of **2a** increased dramatically to 98% (Entry 2). (*E*)-Enyne **2a** was not formed using only the rhenium catalyst, $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$, or TBAF (Entries 3 and 4). In these reactions, **1a** was recovered quantitatively. These results show that a combination of $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ and TBAF is necessary to promote the reaction. In addition, the ratio between $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ and TBAF is important. When using 2.5 mol % of $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ and 5.0 mol % of TBAF, (*E*)-enyne **2a** was formed in only 4% yield (Entry 5). This result is in sharp contrast to Entry 2, in which 10 mol % of TBAF was employed. Other additives were not effective for the reaction (Entries 6–8).

Next, we investigated several terminal alkynes (Table 2). Arylacetylenes bearing electron-donating groups, **1b** and **1c**, provided (*E*)-enynes **2b** and **2c** in excellent to quantitative yields (Entries 1 and 2). By using 1-bromo-4-ethynylbenzene (**1d**), the corresponding (*E*)-enyne **2d** was obtained in 92% yield without loss of bromine (Entry 3). Enyne **2e** was produced in 92% yield when an arylacetylene having an electron-withdrawing group, **1e**, was employed (Entry 4). Arylacetylenes with substituents at either the meta- or ortho-positions, **1f** and **1g**, also afforded (*E*)-enynes **2f** and **2g** in 92 and 94% yields, respectively (Entries 5 and 6). (*E*)-Enyne **2h** was also produced from 2-ethynyl-6-methoxynaphthalene (**1h**) (Entry 7). Enyne **1i** generated trieneyne **2i**, albeit in low yield (Entry 8). However, alkylacetylene, such

Table 2. Investigation of several terminal alkynes **1**

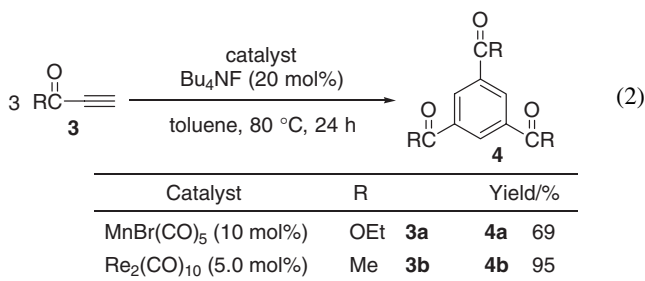
$2 \text{ R}-\text{C}\equiv\text{C}-\text{R} \xrightarrow[\text{toluene, 80 }^\circ\text{C, 24 h}]{[\text{ReBr}(\text{CO})_3(\text{thf})]_2 (5.0 \text{ mol}\%), \text{Bu}_4\text{NF} (20 \text{ mol}\%)} \text{R}-\text{C}\equiv\text{C}-\text{CH}=\text{CH}-\text{R}$		
Entry	R	Yield/% ^a
1	4-(MeO)C ₆ H ₄ 1b	2b 90 (94)
2	4-MeC ₆ H ₄ 1c	2c 98 (>99)
3	4-BrC ₆ H ₄ 1d	2d 92 (96)
4	4-(CF ₃)C ₆ H ₄ 1e	2e 92 (94)
5	3-MeC ₆ H ₄ 1f	2f 92 (98)
6	2-(MeO)C ₆ H ₄ 1g	2g 94 (96)
7	 1h	2h 90 (96)
8 ^b	 1i	2i 35 (38)

^aIsolated yield. Yield determined by ¹H NMR is reported in parentheses. ^bRe₂(CO)₁₀ (5.0 mol %) was used as a catalyst. 115 °C.

as 4-phenyl-1-butyne, and triisopropylsilylacetylene did not produce the corresponding (*E*)-enynes, and those alkynes were recovered quantitatively.

When ethyl propiolate (**3a**) was used as the terminal alkyne, and treated with a catalytic amount of the rhenium complex, [ReBr(CO)₃(thf)₂] (2.5 mol %), and TBAF (10 mol %) in toluene at 80 °C for 24 h, the corresponding (*E*)-enynes was not formed. Instead, the cyclotrimerization of alkyne **3a** proceeded, and the 1,3,5-trisubstituted aromatic compound **4a** was obtained in 60% yield (eq 2).^{7,8} In this reaction, the regioisomeric 1,2,4-trisubstituted aromatic compound was not formed. Benzene-1,3,5-tricarboxylate **4a** was formed in a trace amount in the absence of TBAF, but produced in 12% yield in the presence of TBAF. By using MnBr(CO)₅ as a catalyst, the yield of **4a** improved slightly (eq 2). An alkyne with an electron-withdrawing group, 3-butyne-2-one (**3b**), also afforded the corresponding 1,3,5-trisubstituted aromatic compound **4b** in 95% yield when Re₂(CO)₁₀ was employed as a catalyst (eq 2).

In summary, we have succeeded in rhenium and TBAF-catalyzed regio- and stereoselective synthesis of (*E*)-enynes or 1,3,5-tricarbonyl-substituted benzenes from terminal alkynes.^{9,10} In these reactions, other regio- and stereoisomers were not produced. We hope that the reactions will become useful methods to synthesize (*E*)-enynes and benzene-1,3,5-tricarboxylates.



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- There has been an example of the synthesis of 1,3,5-trisubstituted aromatic compounds by oxidative cyclotrimerization of acrylate. See: K. Tamaso, Y. Hatamoto, S. Sakaguchi, Y. Obora, Y. Ishii, *J. Org. Chem.* **2007**, *72*, 3603.
- We have recently reported on several rhenium-catalyzed transformations using alkynes. See: a) Y. Kuninobu, A. Kawata, K. Takai, *J. Am. Chem. Soc.* **2005**, *127*, 13498. b) Y. Kuninobu, A. Kawata, K. Takai, *Org. Lett.* **2005**, *7*, 4823. c) Y. Kuninobu, A. Kawata, K. Takai, *J. Am. Chem. Soc.* **2006**, *128*, 11368. d) S. Yudha S., Y. Kuninobu, K. Takai, *Org. Lett.* **2007**, *9*, 5609. e) Y. Kuninobu, H. Takata, A. Kawata, K. Takai, *Org. Lett.* **2008**, *10*, 3133. f) Y. Kuninobu, A. Kawata, M. Nishi, H. Takata, K. Takai, *Chem. Commun.* **2008**, 6360. In Ref. 4f, notes and references section, rhenium-catalyzed dimerization of phenylacetylene has already been reported.
- Results for temperature and reaction time are as follows: 80 °C, 3 h: 30%; 50 °C, 8 h: 16%; and 25 °C, 8 h: trace.
- Results for other solvents are as follows: 1,2-dichloroethane, <1%; THF, 79%.
- Results for temperature are as follows: 50 °C, 9% and 25 °C, trace.
- Ethyl propiolate (**3a**) was not recovered because of the polymerization of **3a**.
- The role of TBAF and the reason for the high selectivity is not clear.
- Supporting Information is available electronically on the CSJ-Journal Web site, <http://www.csj.jp/journals/chem-lett/index.html>.